Microstructural effects and kinetics of high temperature oxidation in Nb-Si base alloys

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Report Documentation Page

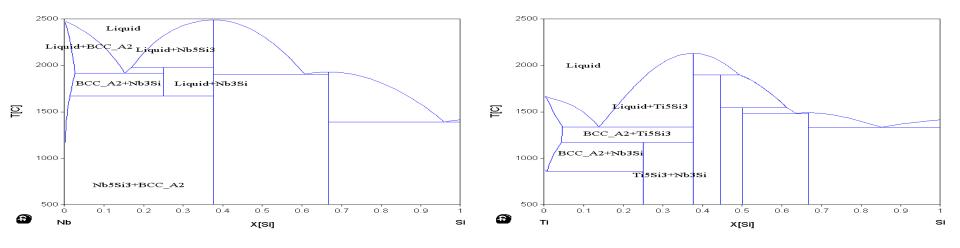
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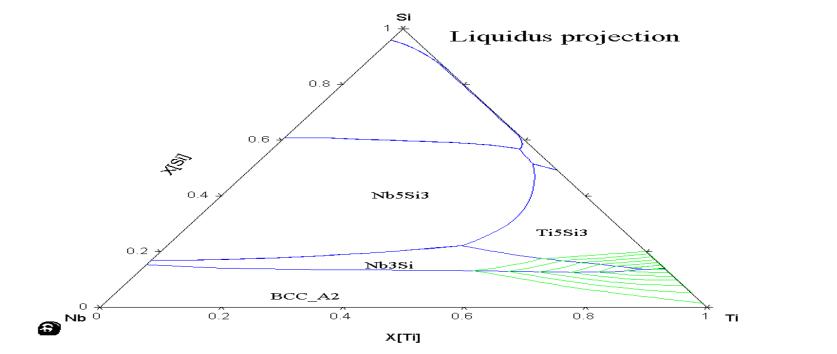
INTRODUCTION

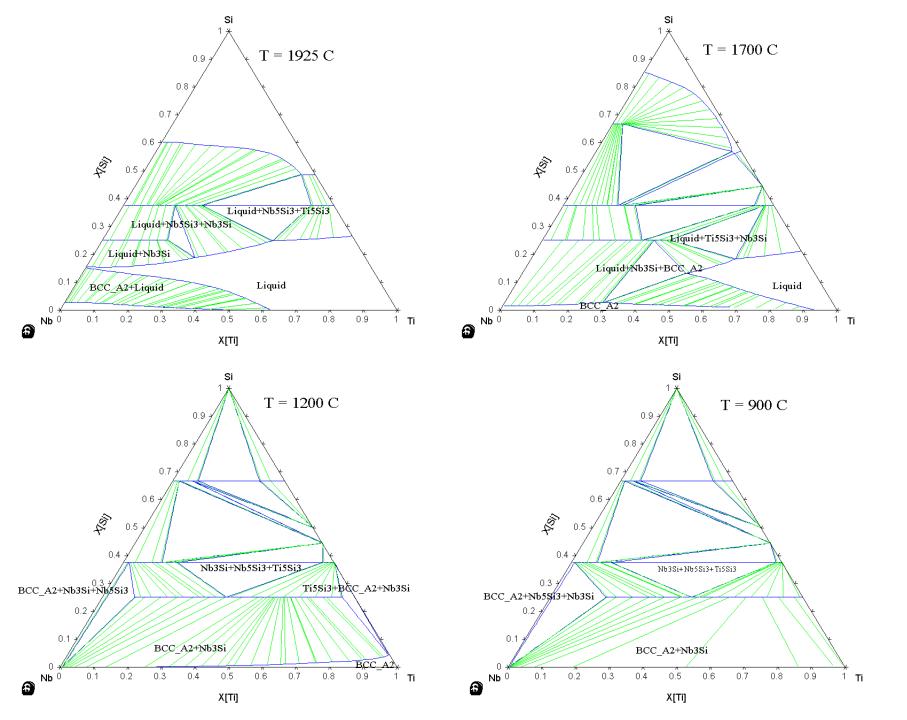
- Understand the effect of alloying on microstructural modification in Nb-Ti-Si based alloys (phases formed, microstructural distribution)
 - Effect of Ti, Al, Cr, C on Nb-Si alloys
 (as-cast, heat-treated alloys)
 - Oxidation effects on microstructures
 - Future directions in the study

Phase diagrams calculated using PANDAT (Y.A. Chang et al.

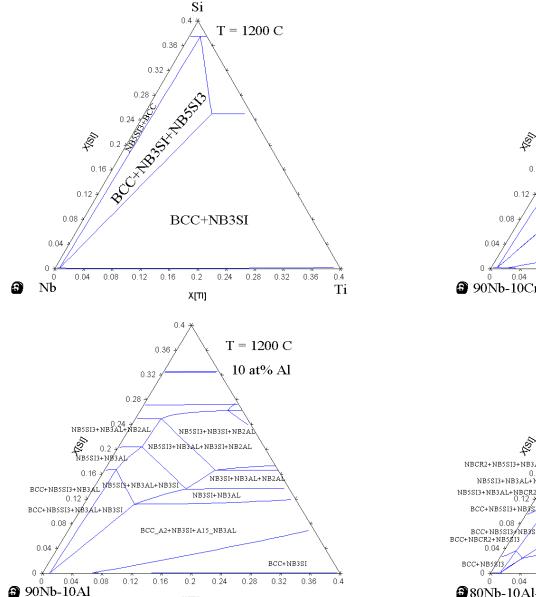
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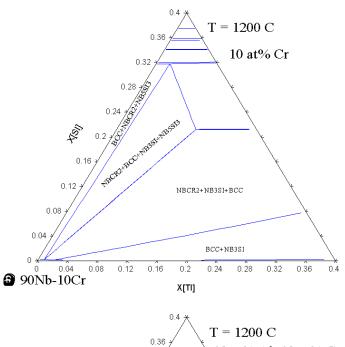


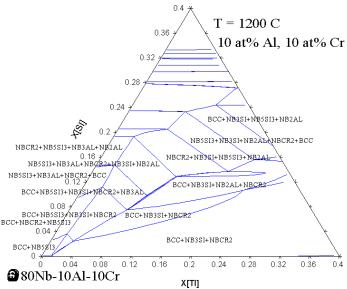


Calculated Isothermal sections at 1200°C

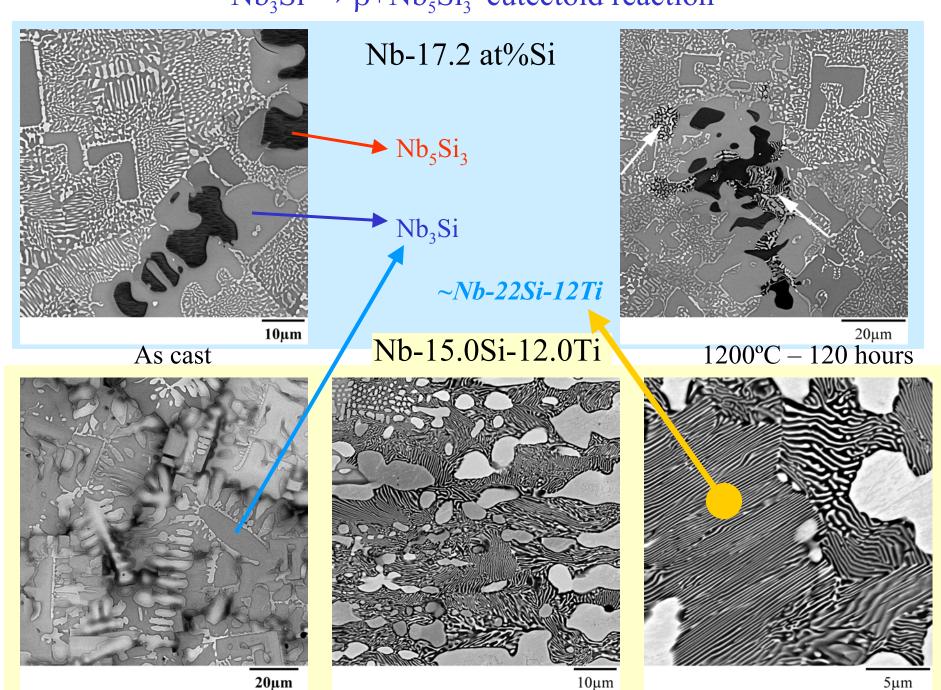


X[TI]

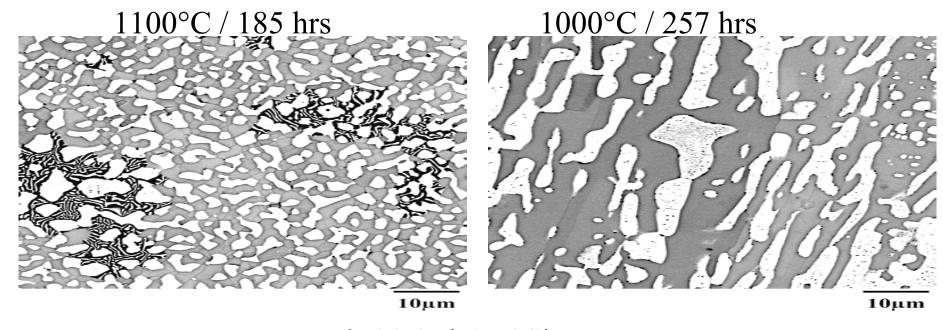




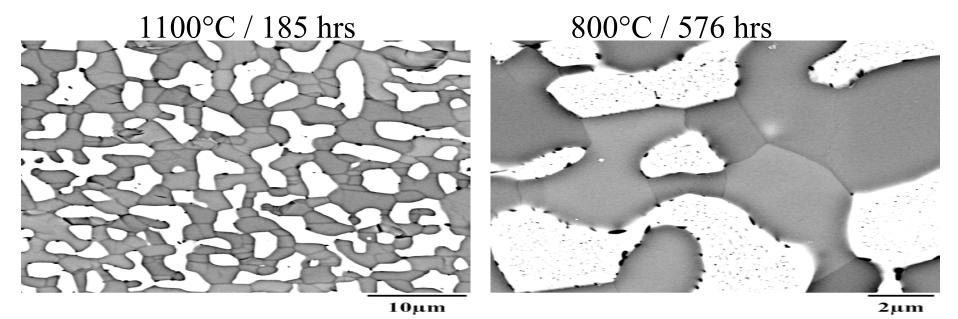
$Nb_3Si \rightarrow \beta + Nb_5Si_3$ eutectoid reaction



Nb-15.8Ti-16.0Si

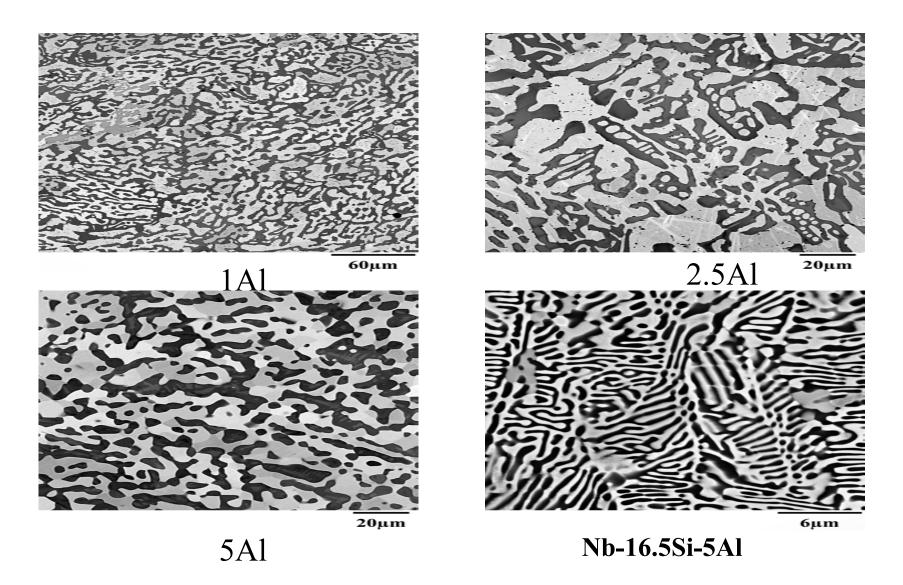


Nb-20.8Ti-15.8Si



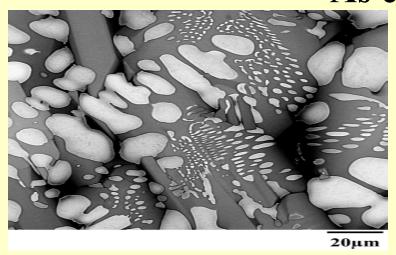
Nb-16.5Si-20Ti-xAl

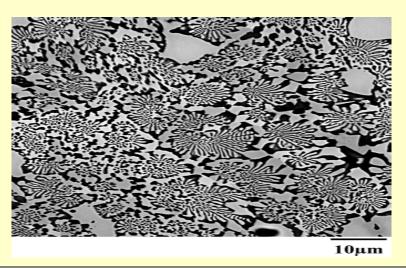
1500°C 100 hours



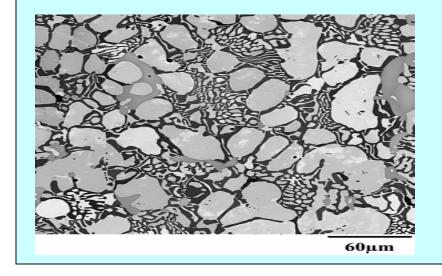
Nb-14Si-12.5Ti-2C

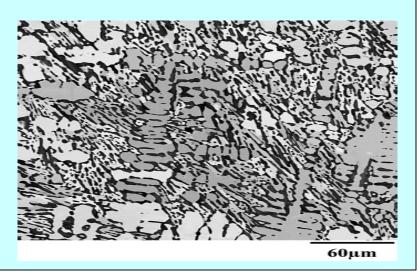
As cast



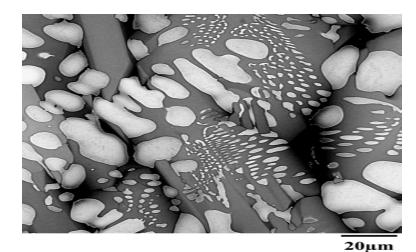


1500°C 100 hours

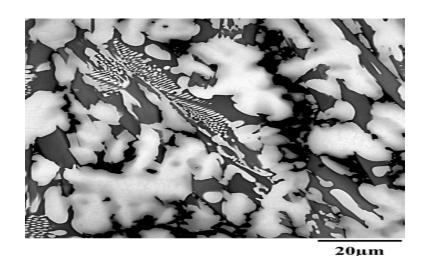




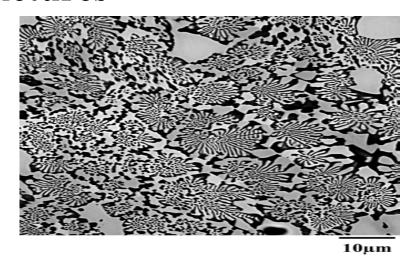
As cast microstructures



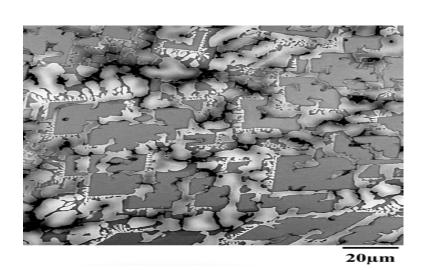
Nb-14Si-12.5Ti



Nb-14Si-12.5Ti-10Cr



Nb-14Si-12.5Ti-2C



Nb-14Si-12.5Ti-10Cr -10Al

Equilibrium Phases and their Compositions at 1200°C

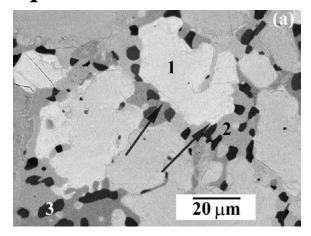


Fig. 1 (a) Nb-19.86Ti-19.74Si-4.21Ge-3.26Al-4.21Hf-9.90Cr

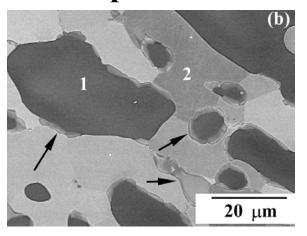
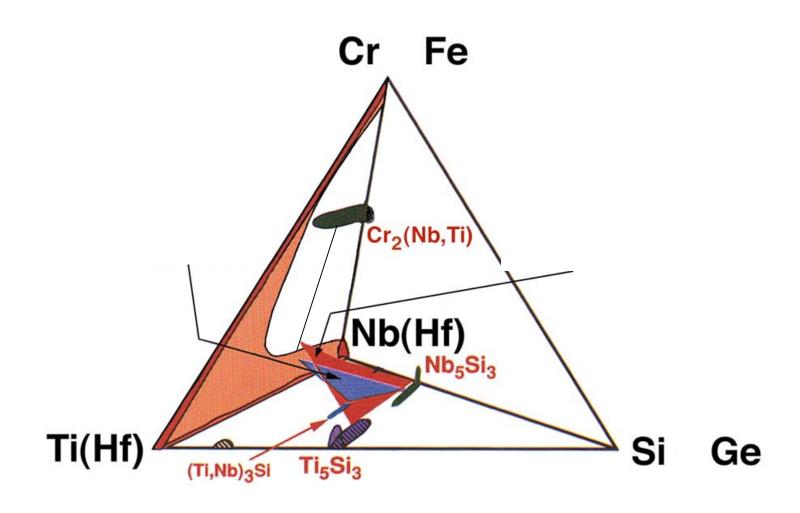


Fig. 1(b) Nb-25.99Ti-12.61Si-4.94Ge-1.92Al-1.90Hf-6.73Cr-0.43Sn.

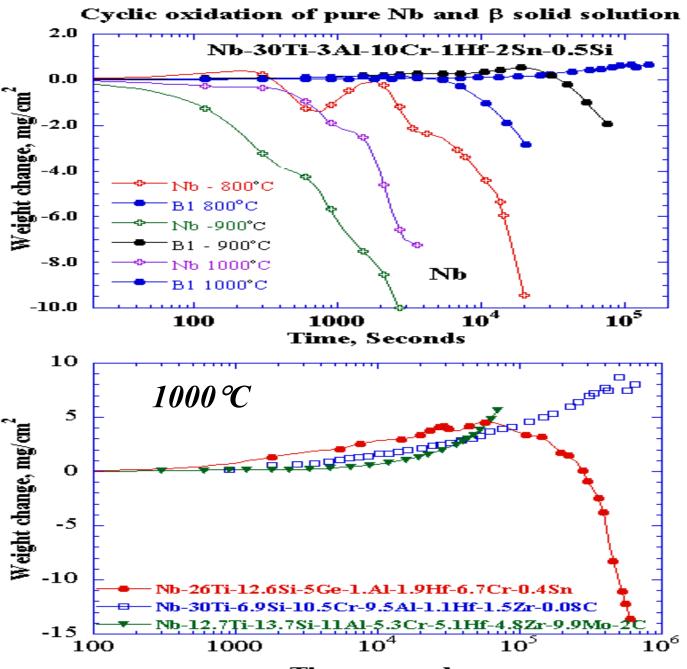
1: Nb₅Si₃; 2: β solid solution phase; 3: Cr₂Nb; Arrows: Ti₅Si₃

	β		Nb ₅ Si ₃ -type		Ti ₅ Si ₃ -type		Cr ₂ Nb -type
At. %	Fig 1(a)	Fig. 1(b)	Fig. 1(a)	Fig. 1(b)	Fig. 1(a)	Fig. 1(b)	Fig. 1(a)
Nb	53.07	57.99	38.04	38.59	28.28	26.80	21.78
Ti	30.23	26.58	18.04	22.23	26.22	29.18	12.65
Si	0.45	0.50	30.83	25.78	27.67	26.47	6.05
Ge	0.16	0.06	5.72	7.34	7.80	9.19	0.34
Hf	1.03	0.72	4.74	1.71	6.73	6.10	4.74
Al	3.35	2.60	1.19	1.54	2.13	1.22	0.98
Cr	11.72	10.06	1.43	2.41	1.17	0.90	54.81
Sn	-	1.51	-	0.41	-	0.14	-

Schematic Phase Diagram

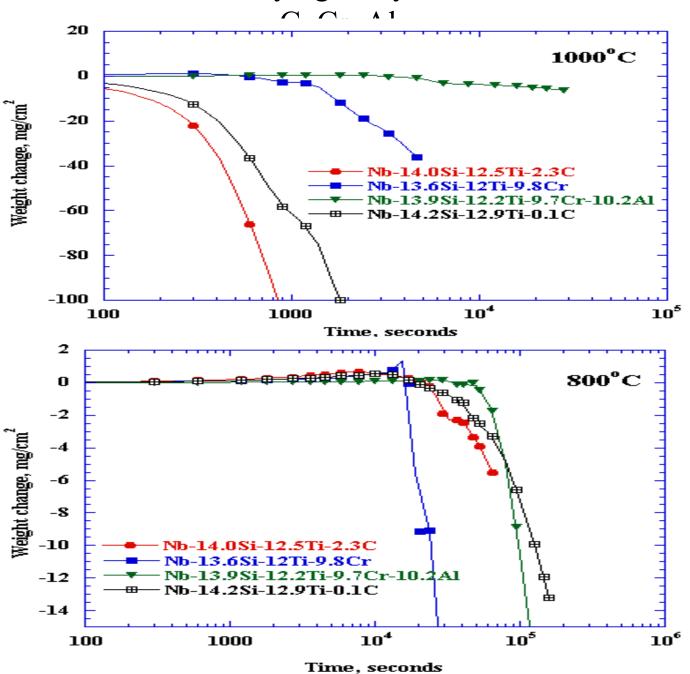


Oxidation resistance of Nb allovs

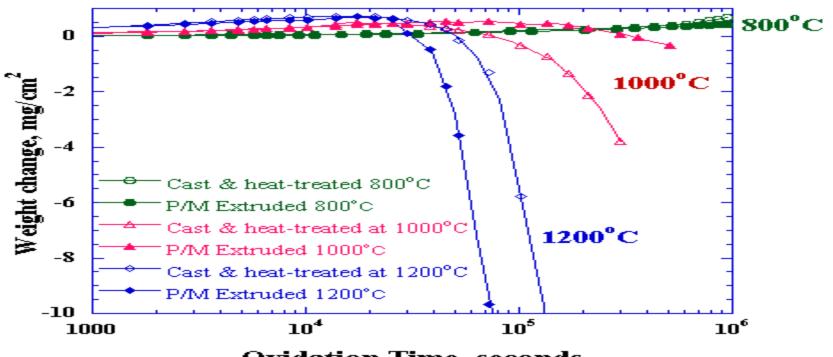


Time, seconds

Effects of alloying on cyclic oxidation



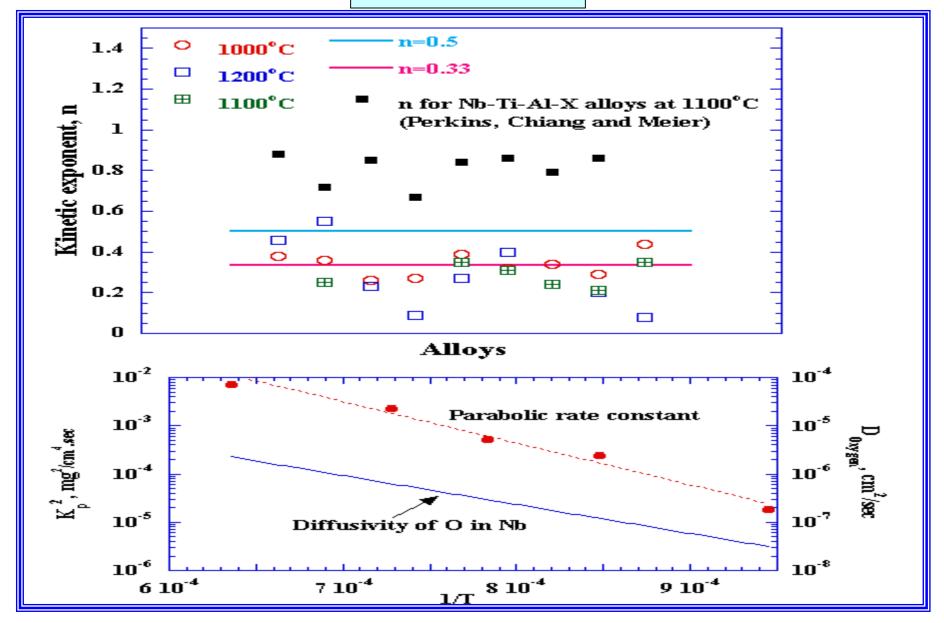
Effect of processing history on cyclic oxidation

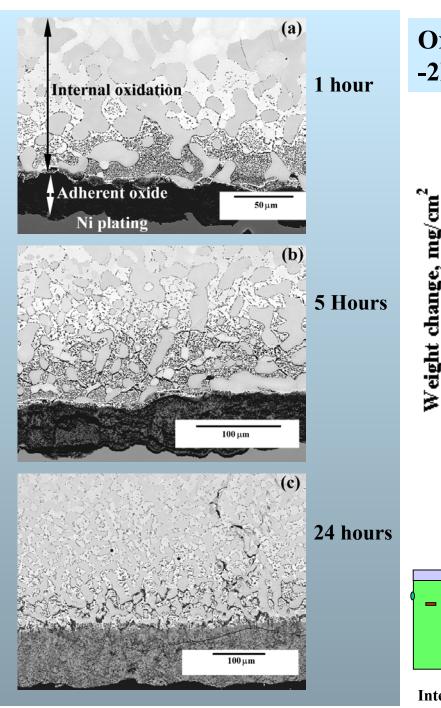


Oxidation Time, seconds

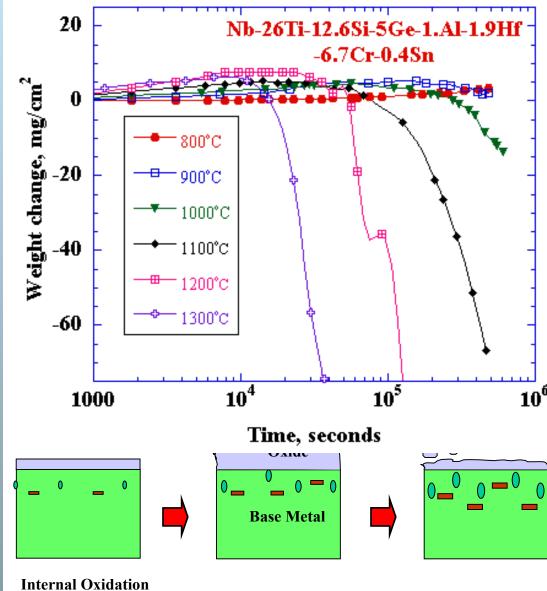
Initial stages of oxidation

$$\Delta w = Kt^{-n}$$

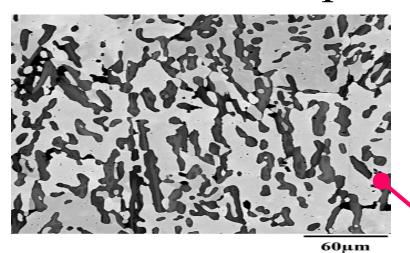




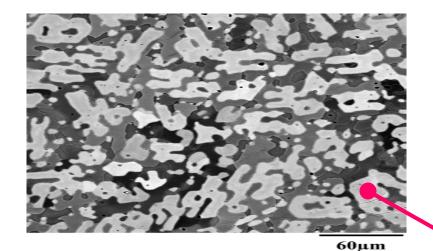
Oxidation of a Nb-26Ti-13Si-5Ge-7Cr-2Al -2Hf-0.5Sn alloy at 1200°C



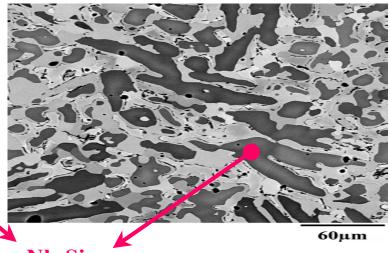
Effect of complex alloying



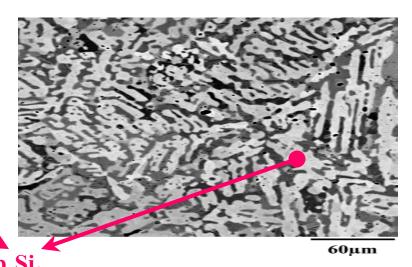
Nb-14Si-12.5Ti-10Cr



Nb-12.5Ti-14Si-10Mo-10Al -5Cr-5Hf-5Zr

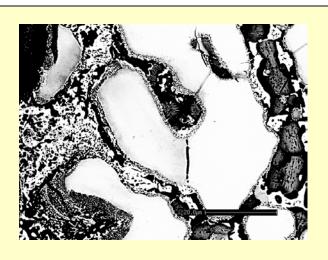


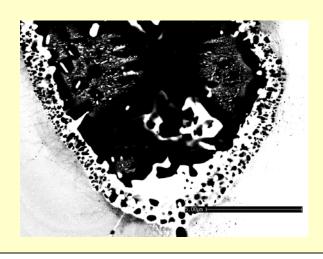
Nb-14Si-12.5Ti-10Cr-10Al



Nb₅Si₃ Nb-12.5Ti-14Si-20Mo-10Al -9Cr-5Hf-5Zr

Preferential oxidation of phases in Nb-20Ti-20Si-4Ge-10Cr-3Al-4Hf-3B Oxidized at 1200°C for 48 hrs

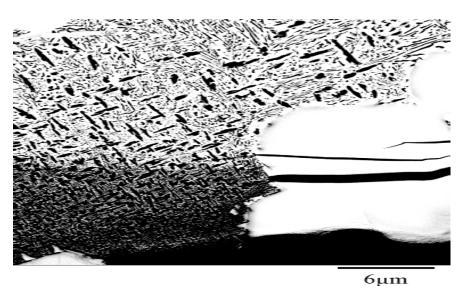




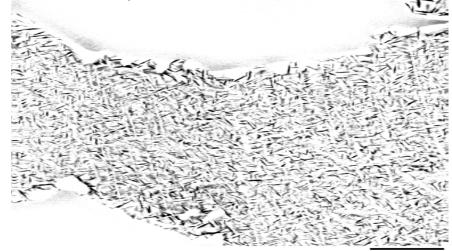
Oxidation of Nb-30Ti-7Si-10.5Cr-9.5Al -1.1Hf-1.5Zr-0.08C



1000 °C for 24 hours

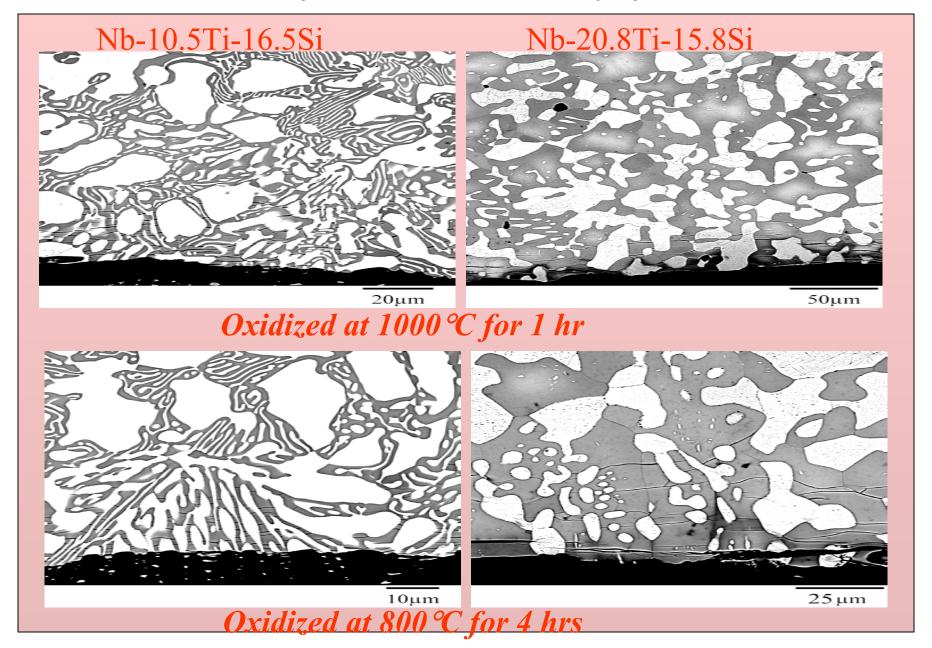


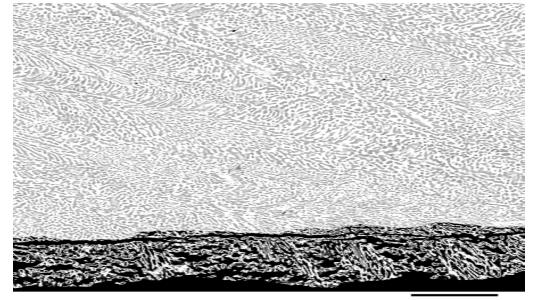
900 °C for 16 hours



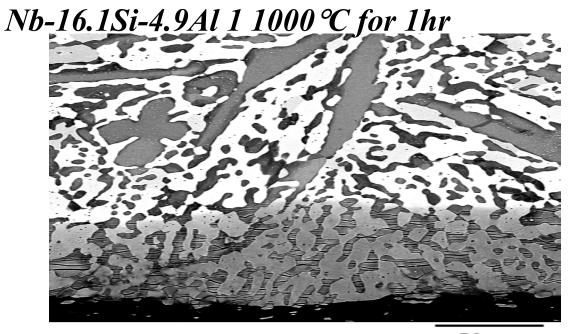
 $2\mu m$

Oxidation of Nb₃Si and lamellar β+Nb₅Si₃



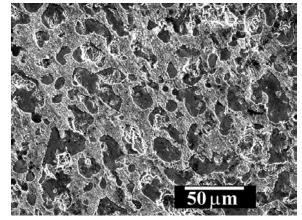


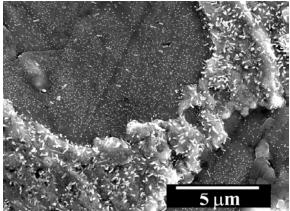
20μm

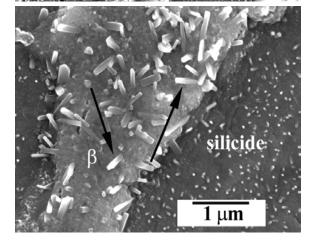


50μm

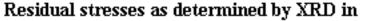
Nb-20.8Ti-15.7Si-4.3Al 800 °C for 4hrs

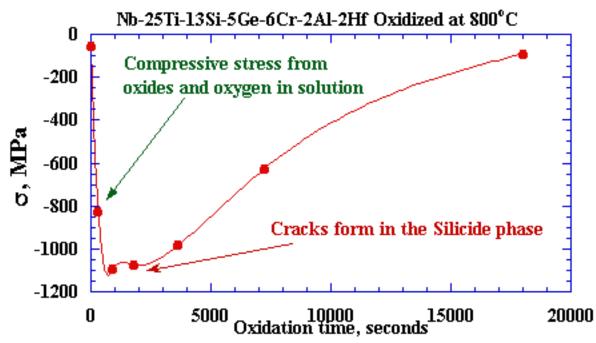




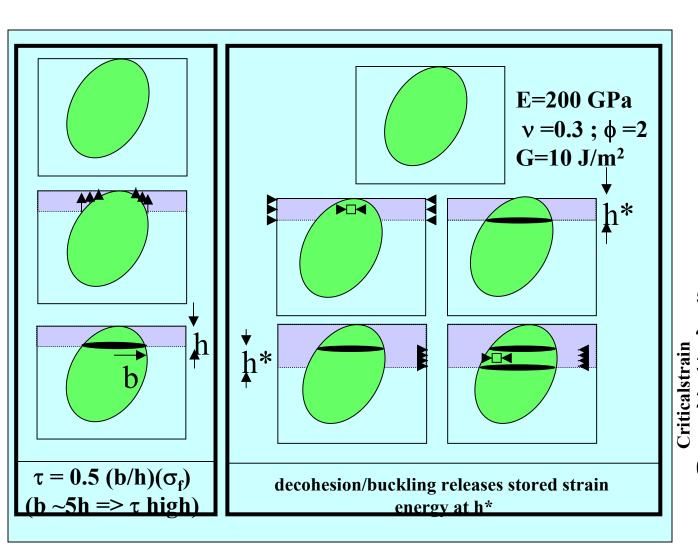


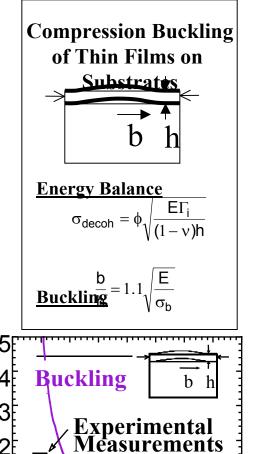
Growth of surface oxide in Nb-26Ti-13Si-5Ge-7Cr-2Al-2Hf-0.5Sn alloy after oxidation at 800°C/5 hrs.





Mechanism of low temperature cracking





Decohesion

10152025303540

b/h (h=5 μ m)

Conclusions

- ➤ It is possible to significantly modify the microstructural distribution of the phases in Nb alloys.
- ➤ Stability of Nb₃Si & Nb₅Si₃ is strongly influenced by alloying additions : Thermodynamic parameters associated with multicomponent systems must be modified.
- > Oxidation resistance of Nb alloys can be increased by alloying.
- > Oxidation behavior is affected by phase distribution in the material.
- ➤ It maybe possible to control the low temperature cracking by microstructural control.